

# Lattice Kinetic Monte Carlo: Building a bridge between *ab-initio* calculations and process simulation

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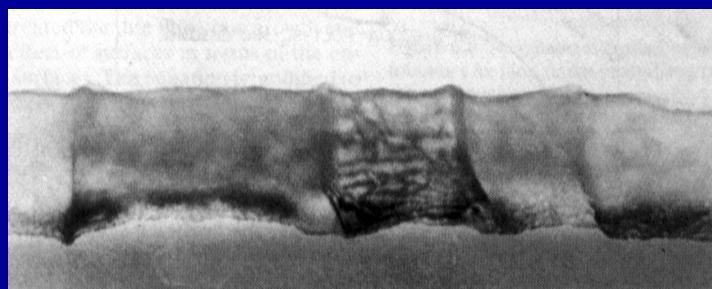
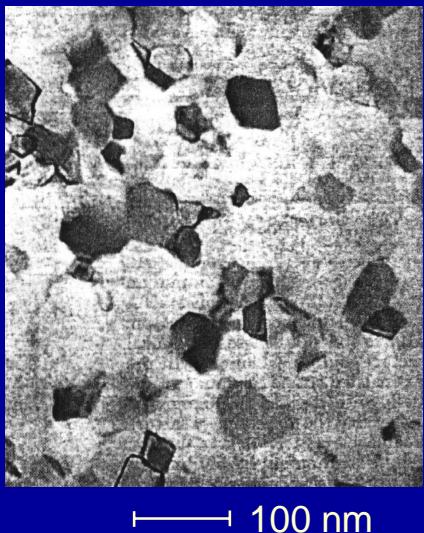
## Polycrystalline Materials

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# Why Lattice KMC?

Aluminum, Polysilicon, ...: Polycrystalline structure



→ 0.5  $\mu$ m

- Kang et al, J. Electron. Mater. , 1997
- Murarka, “Metallization”, 1993

- Surface texture
- Grain Boundary diffusion
- Electromigration

} deposition rate  
depends on: temperature  
substrate

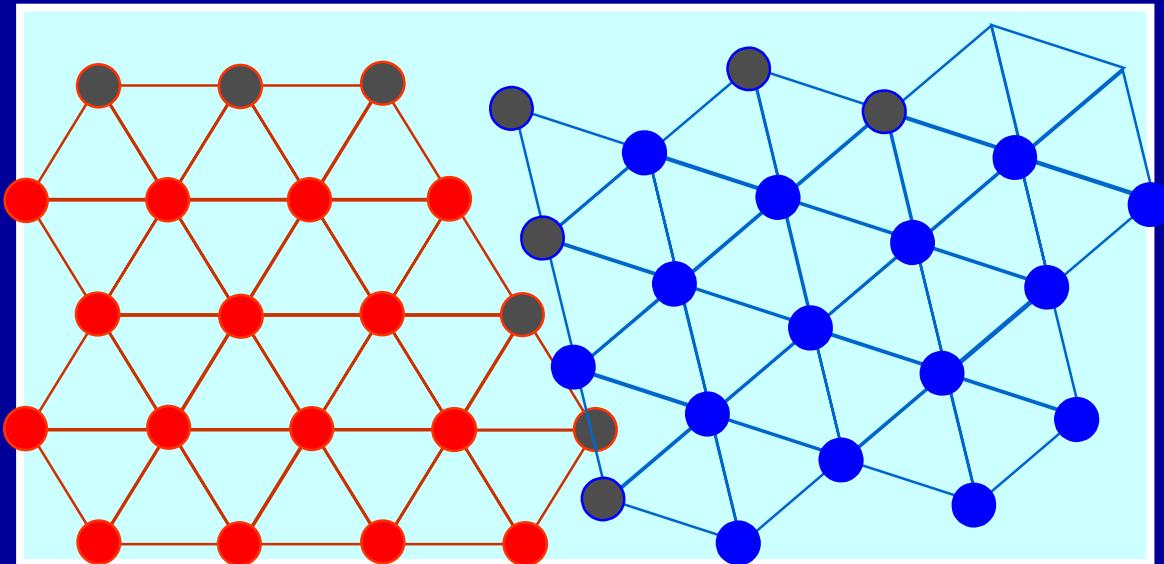
# Simulation approaches

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- ☛ Continuum equations:  
SPEEDIE, SAMPLE, EVOLVE, DEPICT
- ☛ 2-D ballistic simulator:  
SIMBAD, GROFILMS
- ☛ 3-D atomistic approach: ADEPT

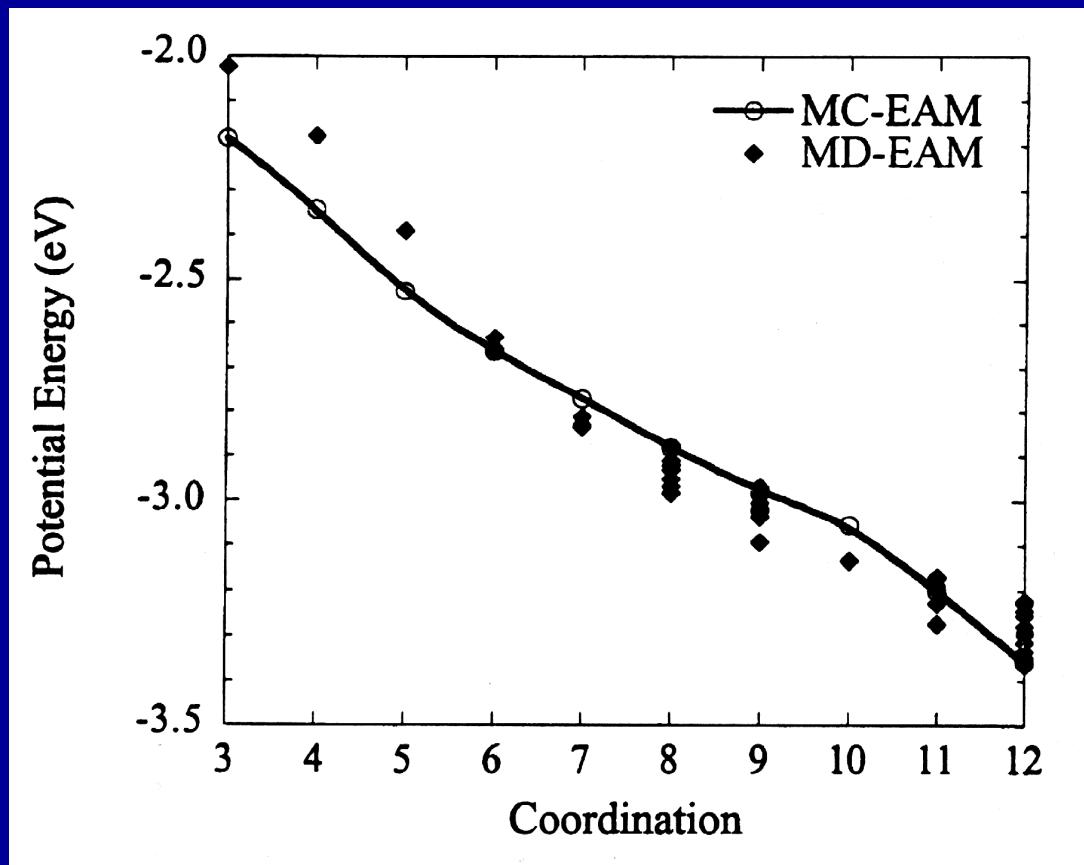
This Work. Polycrystalline 3-D atomistic approach

# Simulation Box

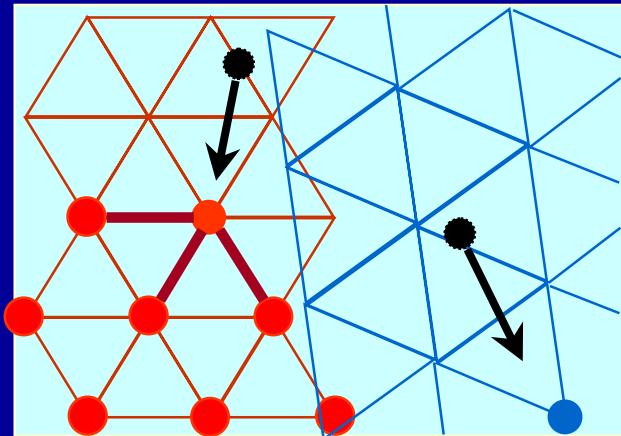
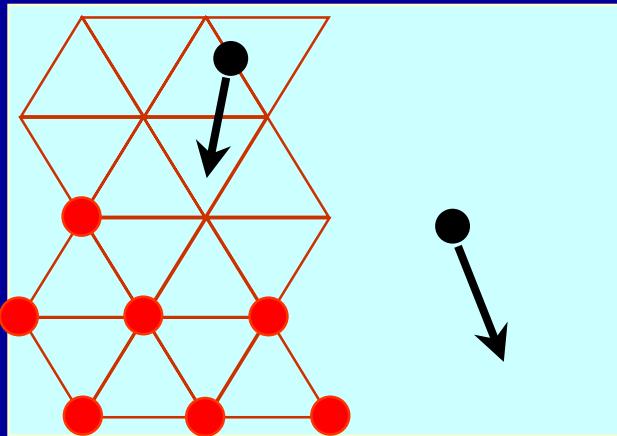


- Atoms are attached to the sites of a perfect fcc lattice.
- As a first approximation, the binding energy of each site depends on the number of occupied nearest neighbors

# Energies from MD Embedded Atom Method using Gupta Potential

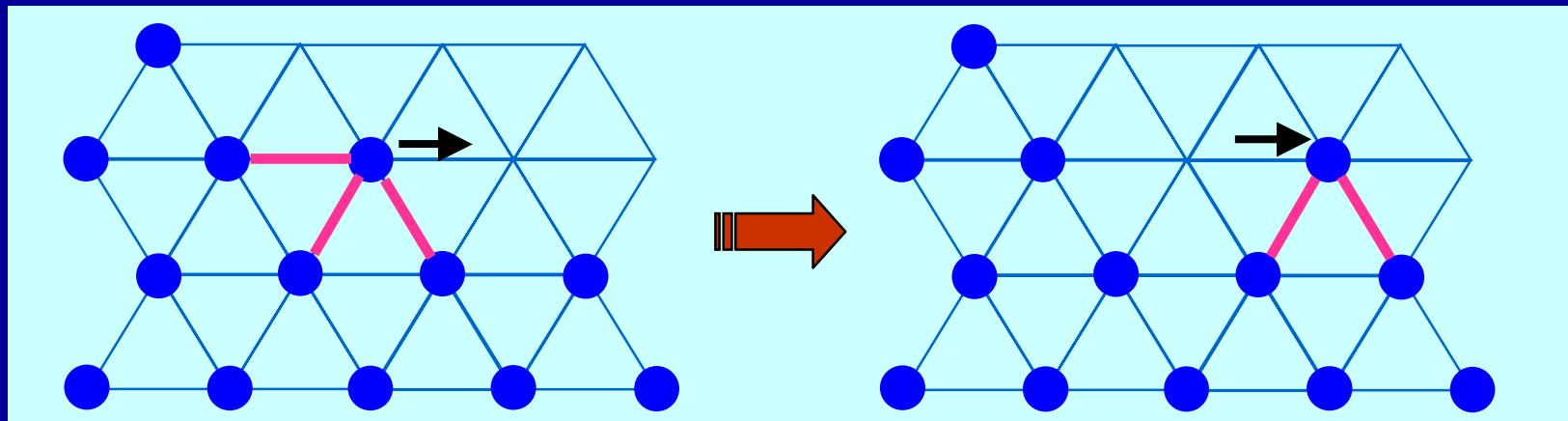


# Deposition



- Initial position and velocity according to desired distribution
- Atom travels in a straight line until it either:
  - Finds some neighbors: gets attached to that grain
  - Finds no neighbors: starts a new grain (orientation).

# Surface Diffusion

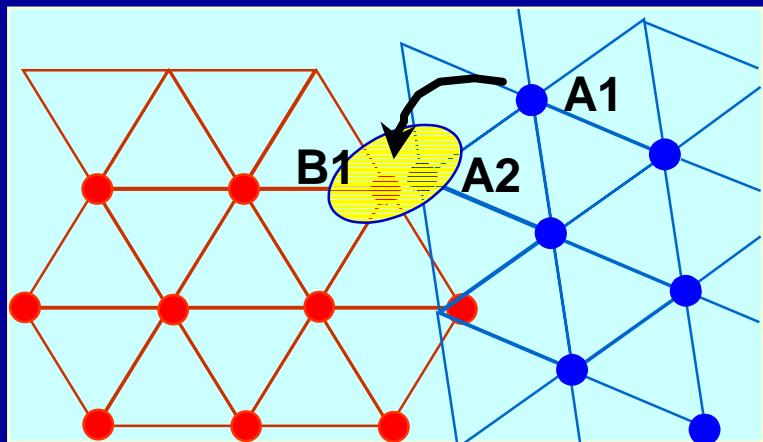
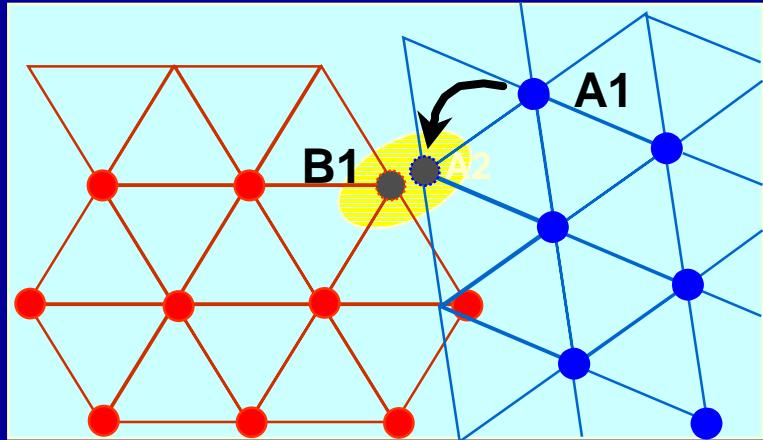


- Atoms not fully coordinated can jump to a neighboring empty site.
- The jump probability depends on the number of nearest neighbors and on the migration energy.

# Grain Boundaries

If the destination site of a jump (A2) is at a grain boundary:

1. Check the energy of sites belonging to other orientations (B) around destination site.
2. If energy (B1) < energy(A2) then the jump final site is B1.

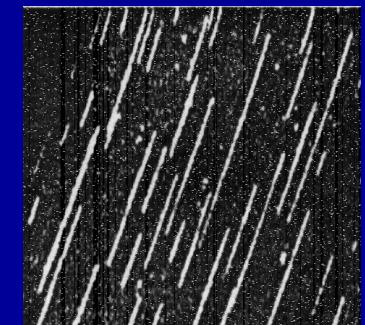
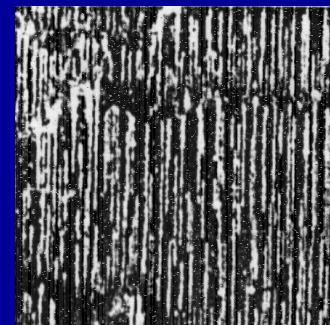
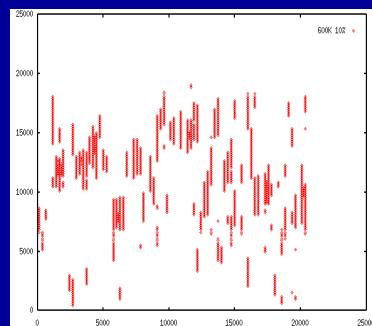
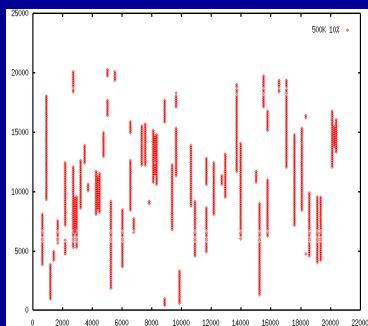
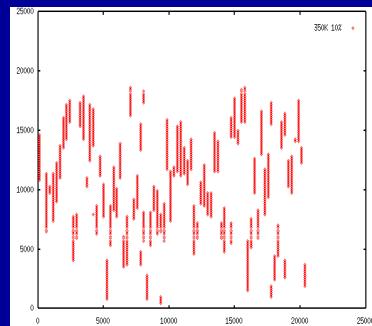
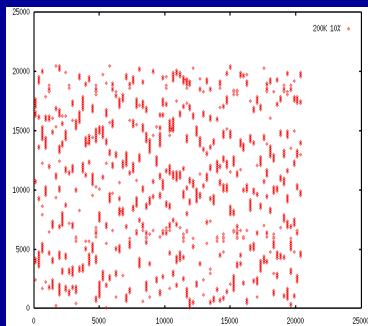


# Lattice KMC: Simulation Examples



- Nucleation
    - Crystalline Substrates: Orientation
    - Amorphous Substrates: Wetting
  - Annealing:
    - Texture evolution
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# Nucleation on Al (110)

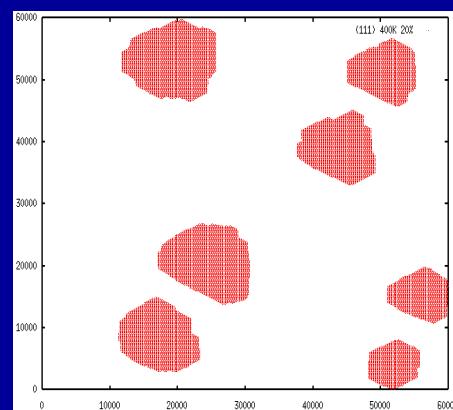
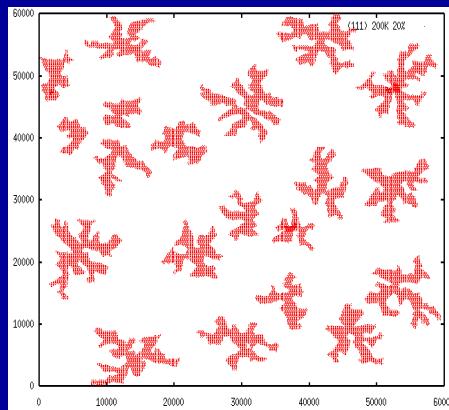


STM images of Cu islands on Pd(100) at 265K and 300 K and coverage of 0.1 ML and 0.07 ML

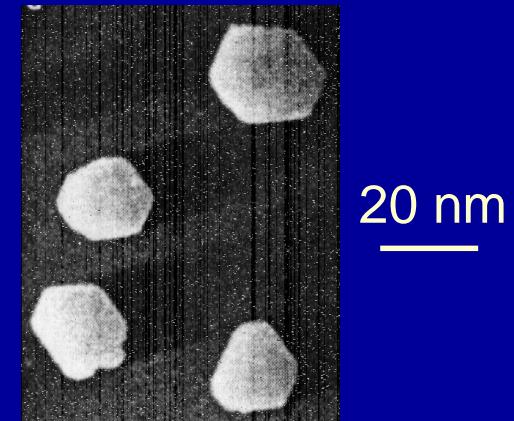
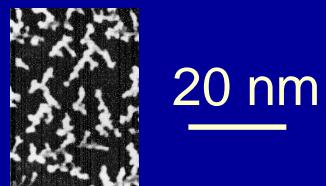
Simulation:  
Al on Al (110) @ 200K, 350K, 500K, 600K  
Coverage=0.1 ML

# Nucleation on Al (111)

Simulation:  
Al on Al (111)  
@ 200K and 400K



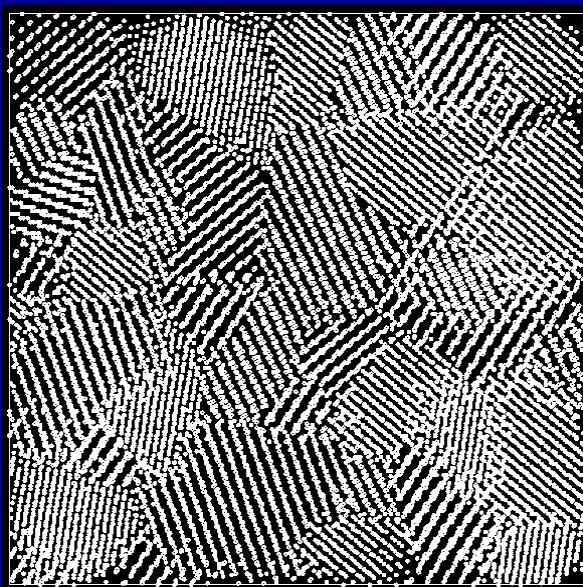
STM measurements:  
Pt on Pt(111)  
@ 200K and 455K



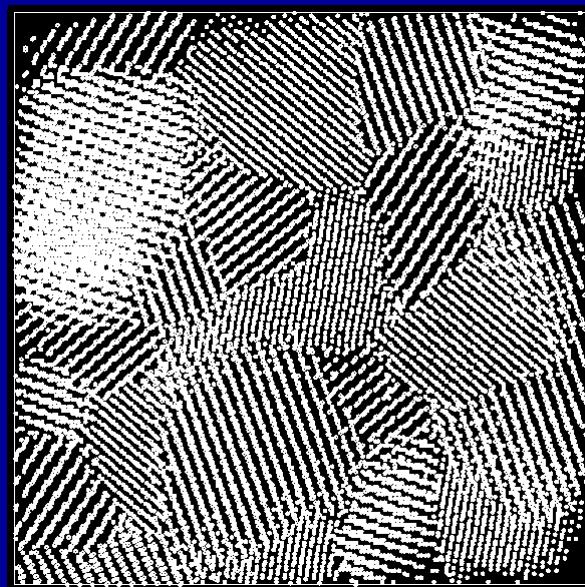
# Nucleation on Amorphous Substrates

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Bonding to substrate: effect on the grain size



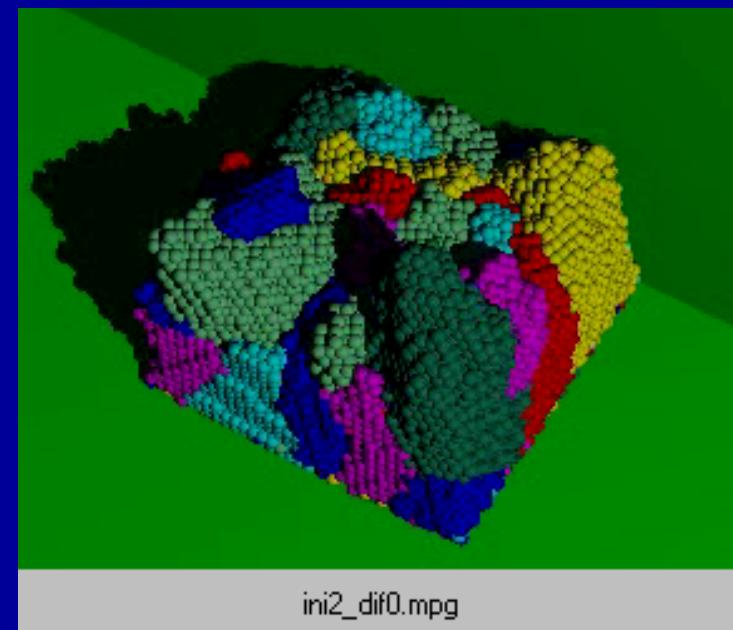
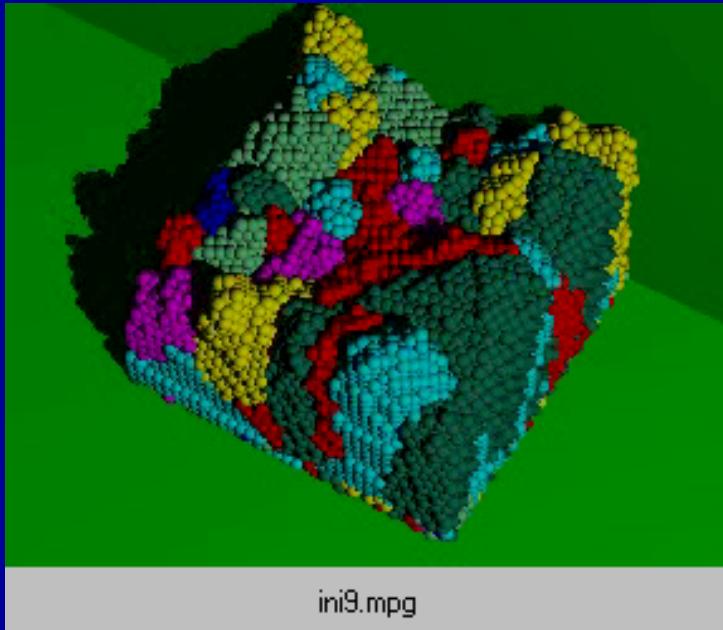
“Wetting” substrate  
(Strong bonding)  
=> smaller grains



“Non-wetting” substrate  
(Weak bonding)

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# Nucleation on Amorphous Substrates

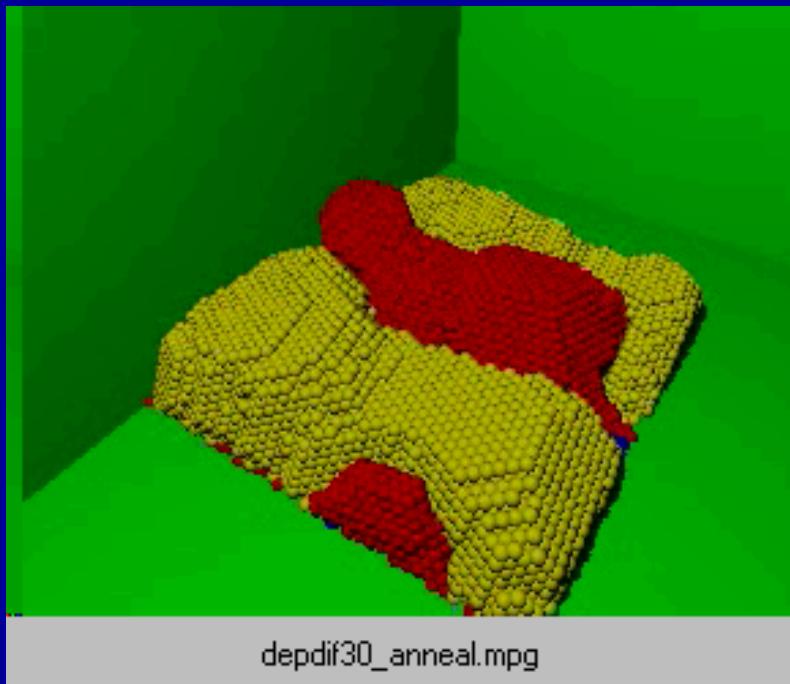


“Wetting substrate”  
(Strong bonding)  
=> smaller grains

“Non wetting substrate”  
(Weak bonding)

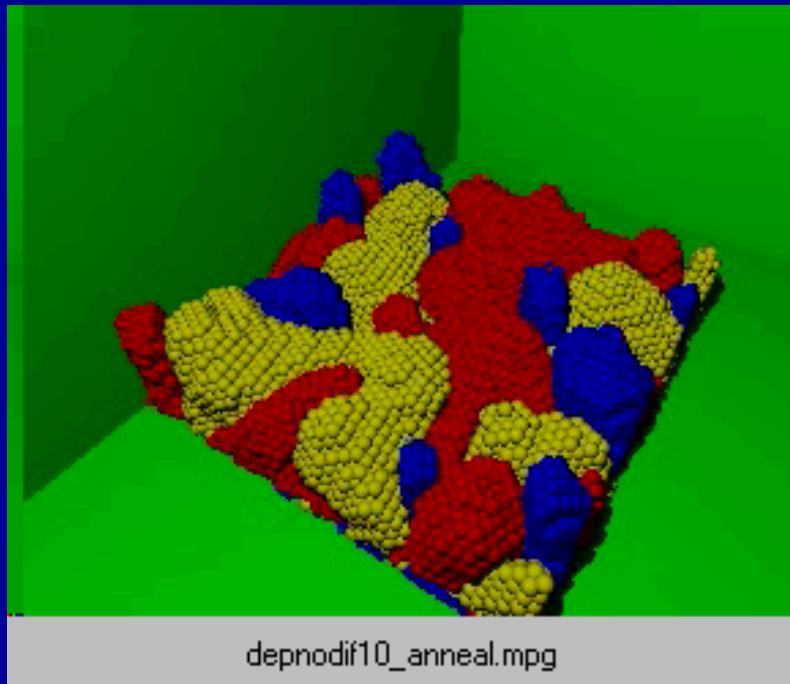
# Annealing: Texture Evolution

$T_{dep} = T_{ann} = 663\text{ K}$   
Annealing time = 2.7 ms



Number of atoms: 20000  
Simulation size:  $0.012\text{ }\mu\text{m}$   
⇒ Actual size:  $0.25\mu\text{m}$

$T_{dep} = T_{ann} = 513\text{ K}$   
Annealing time = 0.18 s



# Conclusions

- Atomistic Process Simulators provide a bridge between ab initio calculations and standard process experimental data.
- Efficient and accurate 3D simulation.
- Straightforward implementation of new mechanisms.
- Microelectronics Device Processing: “Continuum physics models are no longer sufficient below 100 nm. Tools are needed for the physical and chemical processes at an atomic level” (1997 USA National Technology Roadmap for Semiconductors)